

## Unit 8 Study Guide Chemistry Answers

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Unit 8 Study Guide 1. Energy is the ability to do work or produce heat 2. Energy cannot be created or destroyed it can only be transformed . 3. The energy in the universe is constant. 4. A state function changes independent of the pathway. 5. Temperature is a measure of the random motion of the components of a substance. 6. Heat is measured in joules .

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Unit 8 Study Guide Pre-AP Chemistry e, a starting material for urethane plastic foams, consists of C, H, and N. Combustion of such compounds yields 1-120, and N<sub>2</sub> as products. If the combustion of 9.71 g of aniline yields 6.63 g 1-120 and 1.46 g N<sub>2</sub>, what is its empirical formula? q. I . Th mol ma of iline is 93 g/mol. What is its molecular formula?

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Types of Reactions in Chemistry (Unit 8) synthesis. decomposition. single displacement. double displacement. ONE PRODUCT... - simple to complex... - two or more reactions join t.... ONE REACTANT... - complex to simple... - one reactant breaks down in.... 1 COMPOUND + 1 ELEMENT --> 1 NEW COMPOUND + 1 NEW ELEMENT... (CHE....

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Chapter 8 – Unit Conversions 105 Exercise 8.8 - Density Calculations: (Obj 12) a. A graduated cylinder is weighed and found to have a mass of 48.737 g. A sample of hexane, C<sub>6</sub>H<sub>14</sub>, is added to the graduated cylinder, and the total mass is measured as 57.452 g. The volume of the hexane is 13.2 mL. What is the density of hexane? ? g 57.452 48.737 g =

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Chemistry Unit 8 Study Guide Name: \_\_\_\_\_/20 Identify the substances at positions 1 through 4 of the energy level diagram. Choose from the substances listed below.

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### Chemistry Unit 8 Study Guide

Chemistry Unit 8 Rates of Reaction Study Guide Name: th Due Wednesday 17 April in class Identify the substances at positions 1 through 4 of the energy level diagram. Choose from the substances listed below. products 2. 4. reactants intermediates Reaction progress 1. 3. activated complex Com True or False? L 5.

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The mole is a convenient unit for expressing the relative amounts of substances in chemical reactions. The burning of carbon in oxygen can be written with the 2 oxygen atoms bonded in a single O<sub>2</sub> molecule: . The mole is the most common unit used to express the quantity of a chemical substance.

### The Mole Unit - CliffsNotes Study Guides

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Designed for students in Nebo School District, this text covers the Utah State Core Curriculum for chemistry with few additional topics.

Provides answers and explanations to all in-text and end-of-chapter exercises. Also includes summaries of name reactions, functional-group synthesis and reactions, lists of reagents and abbreviations, and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry. This edition now includes all new artwork, expanded in-text problems, summary quizzes approximately every three chapters, more detailed explanations in solutions, and chapter outlines.

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