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Answer Key. 1 mol of a monatomic ideal gas undergoes the following change: $P_A = 3.00 \text{ atm}$, $V_A = 15.0 \text{ L} \rightarrow P_B = 3.00 \text{ atm}$, $V_B = 55.0 \text{ L} \rightarrow P_C = 6.00 \text{ atm}$, $V_C = 55.0 \text{ L}$. Calculate w , q , ΔE and ΔH for the process. $w = -12.2 \text{ kJ}$, $q = 55.5 \text{ kJ}$...

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Activity 1.3.3 Thermodynamics Answer Key. Introduction. Think back to the last time someone complained about a door being left open. What did you notice about the temperature within the room as a result of the open door? In Activity 1.3.3 you will investigate the effects of work, thermal energy, and energy on a system, as in the case of the room with the door left open. Procedure. Answer the ...

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