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**Teaching Transparency The Periodic Table Answers**  
How are nonmetals designated in this periodic table? \_\_\_boxes are tinted yellow\_\_\_ 16. What can be said about the electron configurations of all the elements in a group? \_\_\_their valence electron configurations are identical\_\_\_ The s-, p-, d-, and f-Block Elements . 1. What are the four sections, or blocks, of the periodic table?

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**Teaching Transparency Worksheet The Periodic Table Answers**  
Using the Periodic Table 1. Identify the number of valence electrons in each of the following elements. a. Ne d. Mg f. Cl h. Si 2. Identify the energy level of the valence electrons in each of the following elements. a. c. d. e. H Ar 1 3. Use the periodic table to write the electron configurations (using noble gas notation) for each of the ...

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