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Page 2 Mark Scheme Syllabus Paper IGCSE – May/June 2013 0620 33 © Cambridge International Examinations 2013 1 (a) (i) element cannot be broken into anything simpler [1] by chemical means [1] OR made up of one type of atom only [2] (ii) compound two or more different elements [1] chemically bonded together [1] (iii) mixture

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Page 7 Mark Scheme Syllabus Paper IGCSE – May/June 2013 0620 31 © Cambridge International Examinations 2013 (b) number of moles of HCl =  $0.020 \times 2.20 = 0.044$  [1] number of moles of LiOH =  $0.044$  concentration of LiOH =  $0.044/0.025 = 1.769$  (mol / dm<sup>3</sup>) [1] accept 1.75 to 1.77 need 2 dp correct answer scores = 2 (c) (for LiCl.2H<sub>2</sub>O)

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