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238 Guided Reading and Study Workbook CHAPTER 22, Hydrocarbon Compounds(continued)
14. Circle the letter of each condensed structural formula for pentane. a. C₅H₁₂ b. CH₃CH₂CH₂CH₂CH₃ c. CH₃(CH₂)₃CH₃ d. C—C—C—C—C 15. The IUPAC system uses

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_____ to show the number of carbon atoms

SECTION 22.1 HYDROCARBONS(pages 693–701)

744 Chapter 21 • Hydrocarbons Section 221.11.1 Introduction to Hydrocarbons MAIN Idea Hydrocarbons are carbon-containing organic compounds that provide a source of energy and raw materials. Real-World Reading Link If you have ridden in a car or a bus, you have used hydrocarbons. The gasoline and diesel fuel that are used in cars, trucks, and ...

Chapter 21: Hydrocarbons

The Results for Nomenclature Worksheet 7 Naming Hydrocarbon Answer Key. Free Worksheet. Naming Ionic Compounds Worksheet Answer Key. Problems Worksheet. Triangle Congruence Worksheet 1 Answer Key. Practice Worksheet. Cell Membrane Coloring Worksheet Answer Key. Practice Worksheet.

Nomenclature Worksheet 7 Naming Hydrocarbon Answer Key ...

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Alkanes Worksheet Key 1. Answer the following. a. What is the general formula for a noncyclic alkane? C_nH_{2n+2} b. If a noncyclic contains 15 carbon atoms, how many hydrogen atoms does it contain? 32 2. Draw the line bond structures of following types of hydrocarbons using four carbon atoms: a. two noncyclic alkanes b. two cycloalkanes 3.

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Alkanes Worksheet and Key - Saddleback College

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What are Hydrocarbons? Hydrocarbons are organic compounds containing only hydrogen and carbon atoms. For example, the gas used in gas stoves and bunsen burners can be either propane (C₃H₈) or butane (C₄H₁₀). Combustion of Hydrocarbons All hydrocarbons burn 'cleanly' to form water and carbon dioxide in the presence of sufficient oxygen.

HYDROCARBONS - Qld Science Teachers

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Key Concepts 22.1 Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons

CHAPTER 22 Study Guide

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Select problems and solutions to chapter. A student is asked to write structural formulas for two different hydrocarbons having the molecular formula C_5H_{12} . She writes one formula with all five carbon atoms in a horizontal line and the other with four carbon atoms in a line, with a CH_3 group extending down from the first attached to the third carbon atom.

12.E: Organic Chemistry- Alkanes and Halogenated ...

Day 1: Alkanes - You must know how to name and draw straight chain and branched alkanes. Homework: Finish the Straight Chain Alkane practice at the end of the notes. GOALS: Naming, molecular formula, structural formula, condensed formula, and carbon skeletal formula. I couldn't find another worksheet, but the workbook covers these topics.

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