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===== CHAPTER-14 ===== Chapter 14: Chemical

Equilibrium [1] Which is the correct equilibrium constant

expression for the following reaction? $\text{Fe}_2\text{O}_3(\text{s}) + 3\text{H}_2(\text{g})$

$2\text{Fe}(\text{s}) + 3\text{H}_2\text{O}(\text{g})$ A) $K_c = [\text{Fe}_2\text{O}_3 \dots$ 121 C) 0.0246 D)

0.0179 Ans: D ...

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