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Questions : 1. The number of electrons in Na⁺ (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element which can exhibit covalency as well as electrovalency will be: (a)...

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The four atomic species can be represented as follows. Out of these, the two species which can be termed isobars are: (i) $^{201}_{60}\text{X}$ (ii) $^{200}_{61}\text{X}$ (iii) $^{200}_{58}\text{X}$ (iv) $^{203}_{60}\text{X}$

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Chemistry Chapter 4 Atomic Structure Test

Weighted Atomic mass is $26.4 + 8.9 = \underline{\hspace{2cm}}$ amu. Title: Notes Chapter 4: The Structure of the Atom Author: Administrator Last modified by: Kelli Pearce Created Date: 9/15/2015 10:36:00 PM Company: MISD Other titles: Notes Chapter 4: The Structure of the Atom ...

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