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Answer Key Chapter 19 - Chemistry

2e | OpenStax. 1. (a) Sc: [Ar]4 s² 3 d¹;

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(b) Ti: $[\text{Ar}]4s^2 3d^2$; (c) Cr: $[\text{Ar}]4s^1 3d^5$; (d) Fe: $[\text{Ar}]4s^2 3d^6$; (e) Ru: $[\text{Kr}]5s^2 4d^6$. 3. (a) La: $[\text{Xe}]6s^2 5d^1$, La $3+$: $[\text{Xe}]$; (b) Sm: $[\text{Xe}]6s^2 4f^6$, Sm $3+$: $[\text{Xe}]4f^5$; (c) Lu: $[\text{Xe}]6s^2 4f^{14} 5d^1$, Lu $3+$: $[\text{Xe}]4f^{14}$. 5. Al is used because it is the strongest reducing agent and the only option listed that can provide sufficient driving force to convert La (III) into La.

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2/8 37 4 Marks: 1 Answer: 28 What is the number of electrons? Correct Marks for this submission: 1/1. 5 Marks: 1 Answer: 30 What is the atomic number? Correct Marks for this submission: 1/1. 6 Marks: 1 Choose one answer. a.

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