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absorb only hydrogen and not larger gas molecules, they act simultaneously ...

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Chemistry Chapter 7 Study Guide. the force that holds two atoms together; may form by the attraction of a positive ion for a negative ion or by sharing electrons. a three-dimensional geometric arrangement of particles in which each positive ion is surrounded by negative ions and each negative ion is surrounded by positive ions; vary in shape due to sizes and relative numbers of the ions bonded.

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STUDY GUIDE FOR CHAPTERS 6 & 7. Certain topics like the structure, synthesis, and nomenclature of alkyl halides have already

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been covered previously. This part of the course is primarily concerned with  $S_N1$  and  $S_N2$  reactions,  $E1$  and  $E2$  reactions, and alkene synthesis, structure, and nomenclature. The following is the suggested order of readings and practice problems.

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