

Basic Stoichiometry Phet Lab Answer Key

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[Stoichiometry Lab Report. Stoichiometry I. Introduction/ Purpose: Stoichiometry is the study of the quantitative, or measurable, relationships that exist in chemical formulas and also chemical reactions. The calculations of a stoichiometry problem depend upon balanced chemical equations. The coefficients of the balanced equations indicate the molar ratio of the reactants and products taking part in the reaction.](#)

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[When we bake/cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar.](#)

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grams | Mass of filter paper | 0.50 grams | Mass of coffee filter | 1.00 gram | 150mL beaker + 20mL water + lead nitrate solution | 96.1 grams | 100mL beaker + 20mL water + sodium carbonate solution | 64.2 grams | Watch glass | 32.2 grams | Precipitate + filter paper + coffee filter | 2.20 grams | ...

Results Page 3 About Basic Stoichiometry Phet Lab Answer ...

Stoichiometry Lab Report Chem 121L Part I: Introduction Stoichiometry is the study of the quantitative, or measurable, relationships that exist in chemical formulas and also chemical reactions. In this experiment hydrogen gas will be produced from the reaction of a known mass of magnesium metal with an excess of hydrochloric acid.

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Basic Stoichiometry Phet Lab Homework Exercises Answers

Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. Do the same with chemical reactions. See how many products you can make with different amounts of reactants. Play a game to test your understanding of reactants, products and leftovers. Can you get a perfect score on each level?

Reactants, Products and Leftovers - PhET

Reactants, Products and Leftovers

Reactants, Products and Leftovers

Description. Modified version of "Basic Stoichiometry Lab" by C. Bires. Useful for physical science introduction to limiting reactants. Subject. Chemistry.

Physical Science Limiting Reactants Intro - PhET Contribution

DOC Worksheet for Basic Stoichiometry (ANSWER 386.3g of LiNO₃) 4) Using the following equation: $\text{Fe}_2\text{O}_3 + 3 \text{H}_2 \rightarrow 2 \text{Fe} + 3 \text{H}_2\text{O}$. Calculate how many grams of iron can be made from 16.5 grams of Fe₂O₃ by the following equation. Worksheet for Basic Stoichiometry.

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Worksheet For Basic Stoichiometry Part 1 Answer Key

Basic Stoichiometry Phet Lab Answer Key. Basic Stoichiometry PhET Lab rvsd 2/2011 Let's make some sandwiches! _ Introduction: When we bake/cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar.

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